IN THE CLAIMS:

The listing of claims will replace all prior versions, and listings of claims in the application:

Listing of Claims:

Claim 1(original): "A PROCESS TO OBTAIN TITANIUM CONCENTRATES
WITH HIGH CONTENTS OF TiO₂ AND LOW CONTENTS OF
RADIONUCLIDE ELEMENTS FROM ANATASE MECHANICAL
CONCENTRATES" characterized by the fact that it comprises the following
sequence of unit operations: A process to obtain titanium concentrates with high
contents of TiO₂ and low contents of radionuclide elements from anatase
mechanical concentrates, said process comprising the steps of:

<u>calcinating</u> ore <u>ealeination</u>—in <u>a</u> fluidized bed or rotary kiln, in the <u>a</u> temperature range of 400°C to 550°C, <u>during</u> for 30 minutes to one hour, converting hydrated iron oxides into hematite and <u>promoting less needed reducing</u> time <u>necessary</u> for the <u>a</u> next reduction step, to form a calcinated product;

magnetizing conducting reduction of the calcinated product in the fluidized bed or rotary kiln, at 400°C to 550°C, during for 5 to 30 minutes, using hydrogen,

carbon monoxide, natural gas or any other reducing gas, in order to convert hematite into magnetite, to form a reduced product;

conducting low-intensity magnetic separation of the reduced product; in drum separators, in a 600 to 800 Gauss magnetic field, thereby extracting the magnetic fraction formed in the reduction stage, to form a low-intensity non-magnetic fraction;

conducting a dry, high-intensity magnetic separation of the low-intensity non-magnetic fraction; in a drum or roll separators separator and a rare-earth permanent magnet; in a 16000 to 20000 Gauss magnetic field, thereby extracting the magnetic fraction from silicates, secondary phosphates, monazite, calzirtite, zirconolite and uranium and thorium bearing minerals, to form a high-intensity magnetic fraction;

conducting hydrochloric acid leaching of the high-intensity magnetic fraction, in agitation or column tanks, with 20 to 30% w/w HCl, with-in a 1/2 w/w solid-liquid ratio, at a temperature between 90°C and 107°C, for 2 to 4 hours, thereby solubilizing primary phosphates, iron oxides, aluminium, magnesium, barium and calcium, to form a leached product;

<u>conducting</u> filtration of <u>the</u> leached product; in <u>a</u> belt filter, to form a first filtrated product;

drying of <u>the first filtrated product</u> in <u>a rotary or fluidized-bed drier, to form</u> a first dried product;

oxidation oxidizing of the first dried ore in a rotary kiln or fluidized bed reactor, under a flow of air or oxygen, at a temperature range of 1000°C to 1100°C, in the a presence of a mixture of sodium fluoride (NaF) and amorphous silica (SiO₂), in a 3% to 10% NaF and 1% to 10% SiO₂ proportion with respect to the an amount of material fed to oxidation, thereby forming in the a boundary of anatase grains a radionuclide-rich vitreous phase, in addition to promoting radionuclide migration to the an iron-rich phase, to form an oxidation product;

quenching in water of the oxidation product in water, thereby respectively stabilizing the vitreous and the iron-rich phases, to form a first quenched product;

conducting hydrochloric acid leach-leaching of the first quenched product in agitation or column tanks, with 20 to 30% w/w HCl, with in a 1/2 w/w solid-liquid ratio,—at a 90°C to 107°C temperature range, during for 2 to 4 hours, in the—a presence of sodium fluoride (NaF) or hydrofluoric acid (HF), thereby solubilizing the a radionuclide-rich vitreous phase through the action of generated or added ion fluoride action—ion (F) action, to form a second leached product;

filtering of the second leaching leached product in a belt filter, to form a second filtrated product;

drying of <u>the second</u> filtrated product in <u>a</u>rotary or fluidized bed drier, <u>to</u> form a second dried product;

conducting dry, high-intensity magnetic separation (of the second dried product in a 16000 to 20000 Gauss) magnetic field in a drum or roll separator and rare-earth permanent magnet, thereby separating the an iron containing, radionuclide rich fraction, the non-magnetic fraction becoming the end product concentrate and the magnetic fraction being discarded.

Claim 2 (Currently Amended): The PROCESS TO OBTAIN TITANIUM CONCENTRATES WITH HIGH CONTENTS OF TiO₂ AND LOW CONTENTS OF RADIONUCLIDE ELEMENTS FROM ANATASE MECHANICAL CONCENTRATES process according to claim 1, characterized by the fact that wherein the reduction step is carried out with hydrogen, carbon monoxide, natural gas or any other reducing gas in temperature range of 400°C to 550°C, preferably at 500°C, during 5 to 30 minutes, preferably for 5 minutes.

Claim 3 (Currently Amended): The PROCESS TO OBTAIN TITANIUM CONCENTRATES WITH HIGH CONTENTS OF TiO₂ AND LOW CONTENTS OF RADIONUCLIDE ELEMENTS FROM ANATASE MECHANICAL

<u>wherein</u> the separation of impurities rich in iron, silicates, secondary phosphates, monazite, calzirtite, zirconolite and uranium and thorium containing minerals after the reduction takes place step is carried out through the sequential use of operations of low intensity and high intensity magnetic separations.

Claim 4 (Currently Amended): The PROCESS TO OBTAIN TITANIUM CONCENTRATES WITH HIGH CONTENTS OF TiO₂ AND LOW CONTENTS OF RADIONUCLIDE—ELEMENTS—FROM—ANATASE—MECHANICAL CONCENTRATES process according to claim 3, characterized by the fact that the step—of—wherein the magnetic field used in the dry, high intensity magnetic separation step forming the high-intensity magnetic fraction is done in-a rare earth roll—or permanent magnet separator, with magnetic field intensity ranging from 16000 to 20000 Gauss, preferably 20000 Gauss magnetic field.

Claim 5 (Currently Amended): The PROCESS TO OBTAIN TITANIUM CONCENTRATES WITH HIGH CONTENTS OF TiO₂ AND LOW CONTENTS OF RADIONUCLIDE ELEMENTS FROM ANATASE MECHANICAL CONCENTRATES process according to claim 1, characterized by the fact that

wherein the first hydrochloric acid leaching operation after low intensity and high intensity magnetic separations takes place with step is carried out using a solution containing 20% to 30% w/w HCl, preferably 25%, during 2 to 4 hours, preferably for 4,—hours at a temperature between 90°C and 107°C, preferably—of 105°C, without the addition of air or any other oxidizing agent during leaching.

Claim 6 (Currently Amended): The PROCESS TO OBTAIN TITANIUM CONCENTRATES WITH HIGH CONTENTS OF TiO₂ AND LOW CONTENTS OF RADIONUCLIDE ELEMENTS FROM ANATASE MECHANICAL CONCENTRATES process according to claim 1, characterized by the fact that wherein the oxidation step of the product resulting from the first HCl leaching is carried out in a rotary horizontal kiln or in a fluidized bed, at a temperature between 1000°C and 1100°C, in the presence of a mixture of sodium fluoride (NaF) and amorphous silica (SiO₂), with an amount of 3% to 10% NaF, preferably with 6% to 7% NaF and 1% 3% to 10% 4% SiO₂, preferably from 3% to 4% SiO₂, both with respect to the an amount of ore anatase mechanical concentrates fed into oxidation, under continuous air or oxygen injection, with for a duration of 30 to 120 minutes, preferably 60 minutes.

Claim 7 (Currently Amended): The PROCESS TO OBTAIN TITANIUM CONCENTRATES WITH HIGH CONTENTS OF TiO₂ AND LOW CONTENTS OF RADIONUCLIDE ELEMENTS FROM ANATASE MECHANICAL CONCENTRATES process according to claim 6, characterized by the fact that wherein the fluoride containing compound used in the oxidation step includes one of or more of the following substances: lithium fluoride (LiF), sodium fluoride (NaF), potassium fluoride (KF), magnesium fluoride (MgF₂), calcium fluoride (CaF₂) or ammonium fluoride (NH₄F).

Claim 8 (Currently Amended): The PROCESS TO OBTAIN TITANIUM CONCENTRATES WITH HIGH CONTENTS OF TiO₂ AND LOW CONTENTS OF RADIONUCLIDE ELEMENTS FROM ANATASE MECHANICAL CONCENTRATES process according to claim 1, characterized by the fact that wherein the material resulting from the oxidation step is quenched in water, air or any other cooling means.

Claim 9 (Currently Amended): The PROCESS TO OBTAIN TITANIUM CONCENTRATES WITH HIGH CONTENTS OF TiO₂ AND LOW CONTENTS OF RADIONUCLIDE ELEMENTS FROM ANATASE MECHANICAL

<u>wherein</u> the <u>second</u> hydrochloric acid leaching of the <u>product of oxidation and</u> thermal shock step is carried out with a solution containing from 20% to 30% 25% w/w HCl, preferably 25%, during 2 to for 4 hours, preferably 4, at a temperature between 90°C and 107°C, preferably of 105°C and in the presence of sodium fluoride or hydrofluoric acid, with an amount of ranging from 10 g to 30 g of fluoride ion (F) per liter of leaching solution, preferably 20 g of F per liter.

Claim 10 (Currently Amended): The PROCESS TO OBTAIN TITANIUM CONCENTRATES WITH HIGH CONTENTS OF TiO₂ AND LOW CONTENTS OF RADIONUCLIDE—ELEMENTS—FROM—ANATASE—MECHANICAL CONCENTRATES—process according to claim 9, characterized by the fact that wherein the fluoride containing compound used in the second HCl leaching includes one or more of the following substances: lithium fluoride (LiF), sodium fluoride (NaF), potassium fluoride (KF), magnesium fluoride (MgF₂), calcium fluoride (CaF₂) or ammonium fluoride (NH₄F) or hydrofluoric acid (HF).

Claim 11 (Currently Amended): The PROCESS TO OBTAIN TITANIUM CONCENTRATES WITH HIGH CONTENTS OF TiO₂ AND LOW CONTENTS

OF RADIONUCLIDE ELEMENTS FROM ANATASE MECHANICAL CONCENTRATES process according to claim 1, characterized by the fact that the product resulting from the second hydrochloric acid leaching undergoes magnetic separation through a rare-earth permanent magnet, either roll or drum, with wherein the magnetic field intensity ranging from 16000 to 20000 Gauss, preferably used in the dry, high-intensity magnetic separation step of the second dried product is a 20000 Gauss, the non-magnetic fraction resulting from this separation becoming the final concentrate magnetic field.